

## Chemistry 2.7 AS 91167 Demonstrate understanding of oxidation-reduction

## Practice paper 2

1. Mix potassium dichromate solution (acidified) with hydrogen peroxide solution. Record observations and link to species oxidised and reduced with Oxidation numbers and balanced equations Reduced species: Evidence of ON numbers Oxidised species: Evidence of ON numbers Observation linked to species (name/formula/colour of each reactant/product and overall colour change) **Balanced equations:** (balanced half equation then full equation) 1. Mix potassium iodate (acidified) with iron sulfate solution. Record observations and link to species oxidised and reduced with Oxidation numbers and balanced equations Reduced species: \_\_\_\_\_ Evidence of ON numbers \_\_\_\_\_ Oxidised species: Evidence of ON numbers **Observation linked to species** (name/formula/colour of each reactant/product and overall colour change) **Balanced equations:** (balanced half equation then full equation)